COURSE SYLLABUS

1. Program information

1.1. Institution	Petroleum - Gas University of Ploiesti
1.2. Faculty	Petroleum Refining and Petrochemistry
1.3. Department	IPPPM
1.4. Field of study	Chemical Engineering
1.5.Study cycle	Master
1.6. Study program	Chemical Engineering for Refineries and
	Petrochemistry

2. Course information

2.1.Course title	Ethics and academic integrity			
2.2. Course coordinator	Assist. Prof. Ph.D. Eng. Movileanu Daniela Luminița			
2.3.Laboratory / seminar coordinat	or -			
2.4. Year of study	II			
2.5. Semester *	3			
2.6.Evaluation type	V			
2.7. Course type - formative categor	ory **/ C3/C			
Type of subject matter ***				

^{*}the semester number is in accordance with the curriculum;

3. Total estimated time (teaching hours per semester)

				•			
3.1. Number of hours per week	1	of which: 3.2.	1	3.3.Seminar/laboratory	-	3.4.Project	-
		course					
3.5. Total hours from curriculum	14	of which: 3.6.	14	3.7. Seminar/laboratory	1	3.8. Project	-
		course				-	
3.9. Total hours of individual study (Study of textbook, course support, bibliography, study of textbook, course						106	
support, further reading in the library, on online platforms, preparing seminars/laboratories, homework,							
portfolios and essays)							
3.10. Total hours per semester						120	
3.11. Number of credits					4		

4. Prerequisites (where applicable)

	> graduated bachelor
4.1. Curriculum requirements	basic knowledge of using computer technologies for data acquisition,
	data processing and documentation
4.2. Course requirements:	Course room with video projector
4.3. Seminar/Laboratory	
requirements:	

^{**}fundamental = F0; domain = D1; speciality = S2; complementary = C3

^{***}compulsory = C; optional = O; elective= E

5. Specific competences acquired and learning achievements* outcomes

	by an examining admic verification outcomes		
Professional competences	Learning achievements*		
3. Integrates principles of sustainable development and circular economy	K1 - The student describes advanced concepts of sustainable development applicable in chemical engineering. K2 - The student identifies strategies for reducing, reusing, and valorizing resources. K3 - The student defines performance indicators for sustainable processes. S1 - The student evaluates the environmental impact of chemical processes. S2 - The student proposes technological solutions for pollution reduction and energy efficiency. LO1 - The student makes decisions in accordance with environmental legislation and sustainability principles. LO2 - The student promotes ethical conduct in resource use.		
4.Uses advanced techniques of analysis and quality control	 K1 - The student describes modern methods of instrumental analysis and material characterization. K2 - The student explains principles of validation and calibration of analytical methods. 		
	S1 - The student applies advanced experimental methods for product characterization. S2 - The student uses statistical tools for analytical data interpretation. LO1 - The student takes responsibility for validating and reporting results. LO2 - The student prepares quality reports according to international standards.		
5.Carries out research and innovation activities in chemical engineering	 K1 - The student describes advanced research methodologies in chemical engineering. K2 - The student identifies innovative directions for the development of processes and products. K3 - The student defines methods for experiment design and interpretation. S1 - The student applies experimental and computational methods to obtain original results. S2 - The student writes scientific papers and research projects. LO1 - The student demonstrates autonomy in carrying out research projects. LO2 - The student disseminates results nationally and internationally. 		
Transversal competences	Learning achievements*		
Develops critical thinking and the ability to solve complex problems	K1 - The student identifies reasoning models applicable in interdisciplinary contexts. S1 - The student applies methods of analysis and synthesis to solve complex problems. S2 - The student uses modern tools for decision evaluation and substantiation. LO1 - The student takes responsibility for the proposed solutions and their impact. LO2 - The student demonstrates autonomy in the critical approach of complex situations.		
Communicates effectively orally and in writing in Romanian and in an international language	K1 - The student explains the specialized terminology in Romanian and in a foreign language. S1 - The student drafts reports, presentations, and professional documents.		

	S2 - The student delivers oral presentations and debates in academic and
	professional contexts.
	LO1 - The student takes responsibility for the correct and clear transmission
	of information.
	LO2 - The student demonstrates autonomy in selecting means and
	communication strategies.
Collaborates effectively in	K1 - The student explains the dynamics and roles of members in a
_	
multidisciplinary and	multidisciplinary team.
intercultural teams	S1 - The student actively participates in team activities and contributes to
	achieving common goals.
	S2 - The student uses collaboration and communication management tools.
	LO1 - The student assumes responsibility for their role in the team and
	respects cultural diversity.
	LO2 - The student demonstrates autonomy and initiative in conflict resolution
	and collaboration facilitation.
Demonstrates lifelong learning	K1 - The student explains the principles of responsible use of IT resources.
ability and the use of IT	S1 - The student uses digital platforms and resources for documentation and
resources	learning.
	S2 - The student integrates new information in solving professional tasks.
	LO2 - The student demonstrates autonomy in selecting and using learning
	resources.
Displays social responsibility,	K1 - The student describes the principles of professional ethics and social
professional ethics, and civic	responsibility.
spirit	K2 - The student explains the ethical implications of professional decisions.
	S1 - The student applies ethical principles in professional and academic
	activities.
	LO1 - The student takes responsibility for the ethical consequences of
	decisions.
	LO2 - The student demonstrates autonomy in promoting ethical and civic
	• 1
	conduct.

^{*} K – knowledge; S – skills; LO – responsibility and autonomy.

6. Course objectives (based on the competence grid)

6.1. General objective	knowledge and understanding of the issues, concepts and principles of ethics and academic integrity; developing documentation skills; developing skills for understanding and analysis of technical and scientific documents (patents, scientific papers) of various degrees of difficulty.	
6.2. Specific objectives	 knowledge and application of the principles and norms of professional ethics and deontology; identification and analysis of academic ethics and integrity issues, use and citation of sources, objective presentation of data; understanding the importance of performing replicable and reliable research and appreciation of factors that lead to rigorous research; application of critical thinking in solving ethical problems; expression of a responsible attitude towards the scientific field to optimal and creative capitalization of their own potential; teamwork, interpersonal communication and the assumption of specific roles. 	

7. Contents

7.1. Course	Time	Teaching methods	Comments
Ethics, deontology and integrity – introductory notices,	1		
terminology, history. The role of ethics and integrity in the			
academic field			
Intellectual fraud: terminology, legal regulations. Plagiarism.	2		
Authorship, ownership and plagiarism in the digital age. Scientific			
communication and deontology.			
Relations in the community and with the society; students'	1		
behaviour			
The code of academic ethics and deontology. Regulation of	2	Lecture,	
organization and functioning of the university ethics committee		conversation and	
Documentation in scientific research. Ethical challenges caused by	1	debate	
fast development of mass media.			
Data acquisition, management and sharing; Sloppiness vs	1		
Fabrication			
Industrial property. Right protection systems: patent, utility model,	2		
design, trademark. International treaties in patents field.			
Online database of patents and scientific papers	3		
Management of patenting. Legal exploitation of patents. Rights and	1		
obligations			
D'11' 1			_

Bibliography

- 1. Constantinescu, M., Mureșan, V., *Institutionalizarea eticii: mecanisme si instrumente*, Editura Universitatii din Bucuresti, 2013
- 2. *** Ullmann's Encyclopedia of Industrial Chemistry, 40 Volume Set, 7th Edition. Wiley-VCH (Editor), 2011
- 3. ***Kirk-Othmer Encyclopedia of Chemical Technology Fourth Edition, John Wiley & Sons, 1998;
- 4. Erhan, V., Brevetul de Inventie Obtinere si exploatare, Editura Lumina Lex, Bucuresti, 1995
- 5. Sutherland Smith, W., *Plagiarism. The internet and student learning. Improving Academic Integrity*, Routledge, Taylor and Francis Group, New York and London, 2008
- 6. Macfarlane, B., *Researching with integrity. The ethics of academic enquiry*, Routledge, Taylor and Francis Group, New York and London, 2009
- $7.\ Brennecke, P., A cademic\ integrity\ at\ the\ Massachusetts\ Institute\ of\ Technology.\ A\ handbook\ for\ students, 2012$
- 8. Bretag, T., Handbook of Academic Integrity, Springer Reference, Singapore, 2016
- 9. Eaton, S.E., Second handbook of Academic Integrity, Springer International Handbooks of Education, Springer Nature Reference, Switzerland AG, 2024

7.2. Seminar / laboratory	Time	Teaching methods	Comments
Bibliography			
7.3. Project	Time	Teaching methods	Comments
Bibliography			

8. Correlation of the course contents with the demands of the epistemic community representatives, professional associations and representative employers in the field of the program

The content of the course and seminars is in agreement with the curricula of other universities, from our country or abroad. In order to better adapt the curriculum content to the requirements of labour market, meetings with economic partners, graduates and teachers from faculties in chemical engineering field were held.

9.Evaluation

Activity	9.1. Evaluation criteria	9.2. Evaluation methods	9.3. Percentage of final grade
9.4. Course	Correctness and completeness of acquired knowledge The degree of acquiring the specific language	Oral exam with theoretical questions	80%

	Scientific report related to ethics and academic integrity	Power point presentation	20%		
9.5. Seminar /					
laboratory / project					
9.6. Minimum performance standard					
Minimum 5 for each examination subject					

Signature/date Course coordinator Laboratory coordinator Project coordinator 22.09.2025

Date of approval in the department

Head of department

Dean

26.09.2025